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Modelling CO₂ Sequestration in Deep Saline Aquifers

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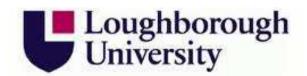
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Geological CO₂ sequestration as one CCS techniques occurs in four mechanisms; hydrodynamic trapping is storage of free CO₂ in pore spaces of sedimentary layers, residual trapping occurs as a result of a hysteresis effect in the permeability of the supercritical phase of carbon dioxide which can happen when the saturation direction is reversed, solubility trapping occurs via CO₂ dissolution in the brine, mineral trapping is a result of carbon dioxide gas dissolves into fluid in the aquifer and reacts with water to form carbonic acid which reacts with dissolved ions within the brine aquifer and minerals forming the host.

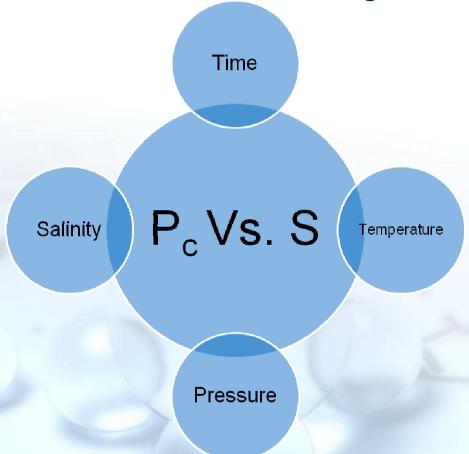
This research project aims to carry out a series of modelling to investigate the injection and transport behaviour of CO_2 in deep saline aquifers as a multiphase flow in porous media in addition to studying the influence of different parameters such as time scale, temperature, pressure, permeability and geochemical condition on the stability of CO_2 underground.

To determine the optimum conditions for the sequestration process which leads to minimize the risk of CO₂ leakage back to the atmosphere. A simulation code named STOMP (Subsurface Transport Over Multiple Phases) is utilized to numerically predict thermal and geological flow and transport phenomena in variably saturated subsurface environments by numerically solving the governing partial differential equations of mass, momentum and thermal energy that describe the subsurface environment simultaneously using Newton-Raphson iterations to resolve the nonlinearities in them. In contrast to most other works which are focussed on determining mass fraction of CO₂, this project focuses on determining capillary pressures of CO₂ as a function of water saturation and CO₂ dissolution rate in groundwater at different time scales, temperature and pressure conditions. A series of figures were produced to illustrate how saturation, capillary pressure and dissolved CO₂ changes with the change of injection process, hydrostatic pressure and geothermal gradient.





Project Aims

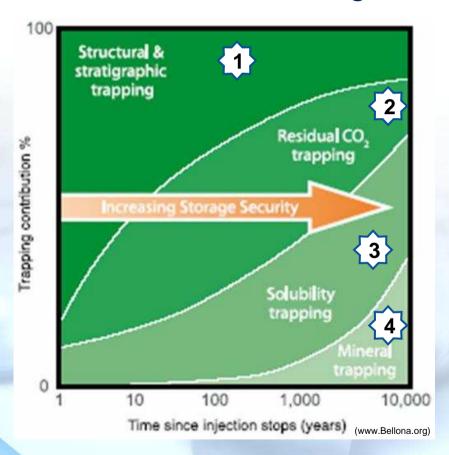


Develop a framework model to determine static and dynamic capillary pressure for water-co₂ system as a function of saturation at various times, temperatures, pressures and salinity.



CO₂ Sequestration in Saline Aquifers

Mechanisms of Storage



- Physical trapping below an impermeable layer
- 2. CO₂ adhered to the solid surface of the formation.
- 3. CO₂ dissolved in the existing fluids and increases it's density or react with information.
- 4. Carbonation reaction with the minerals contained in the formation aqueous.



Modelling CO₂ Injection

Effective Parameters

- Gas composition and injection rate
- SO₂ contaminant CO₂ reduces mineral precipitation and dissolution
- Increasing injection rate increases capillary residual trapping
- Permeability and heterogeneity
- High permeability increases CO₂ mobility
- increased heterogeneity enhances lateral migration and CO₂ dissolution
- Mineral precipitation and sorption
 - Related to the porosity and permeability of the porous media.

Injection Rate

Permeability and Heterogeneity

Modelling

CO2 Injection

Precipitation and Dissolution Sorption Isotherm

gas

composition



STOMP Simulator

Subsurface and Transport Over Multiple Phases



- Competent at analyzing all mechanisms of CO₂ sequestration
- 3. Needs a FORTRAN compiler to generate executable code from source files
- 4. Produces numerical predictions of thermal and hydrogeologic flow and transport phenomena in variably saturated subsurface environments.





STOMP Code Governing Equations

Equations of Mass and Momentum

1. The conservation of motion is governed by Darcy's equation

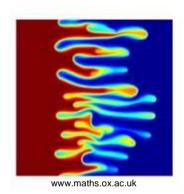
$$\frac{\partial (S_{\alpha} \phi \rho_{\alpha})}{\partial t} + \nabla \cdot (\rho_{\alpha} v_{\alpha}) - \rho_{\alpha} q_{\alpha} = 0$$

2. From the generalized Darcy's law, velocity vector v_{α} can be calculated

$$v_{\alpha} = -\frac{k_{r\alpha}}{\mu_{\alpha}} K \left(\nabla p_{\alpha} - \rho_{\alpha} g \right)$$

3. Solving the 2 equations gives;

$$\frac{\partial (S_{\alpha} \phi \rho_{\alpha})}{\partial t} - \nabla \cdot \left(\rho_{\alpha} \frac{k_{r\alpha}}{\mu_{\alpha}} K (\nabla p_{\alpha} - \rho_{\alpha} g) \right) - \rho_{\alpha} q_{\alpha} = 0$$





STOMP Constitutive relationships

capillary pressure-saturation and relative permeability-saturation

1. Brooks and Corey relationship defines the effective saturation as

$$S_e(p_c) = \frac{Sw - Swr}{1 - Swr} = \left(\frac{p_d}{p_c}\right)^{\lambda}$$
 for $p_c \ge p_d$

2. Brooks-Corey relationships with the Burdine theorem are used to define the relative permeabilitysaturation relationships for wetting and non-wetting phases

$$k_{rw} = S_e^{\frac{2+3\lambda}{\lambda}}$$

$$k_{rnw} = (1 - S_e)^2 (1 - S_e^{\frac{2+\lambda}{\lambda}})$$

λ : form parameter - describes the uniformity of the material
 - usually between 0.2 - 0.3



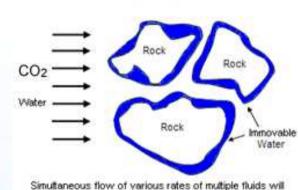
Water-wet rock

Oil is immobile

STOMP Simulation

Assumptions to Simplify Coupled Equations Solution

- 1. Rigid rock (No permeability)
- 2. Both fluids are incompressible
- 3. Constant dynamic viscosity
- 4. Formation is homogeneous
- 5. Isotropic under isothermal conditions
- 6. Thermodynamic equilibrium
- 7. No NAPL phase and no dissolved oil



Multi-Phase Flow

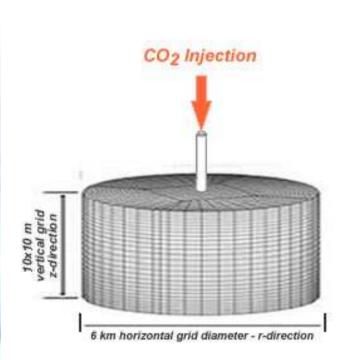
Simultaneous flow of various rates of multiple fluids will result in changing effective permeability to flow for each phase as saturations change.

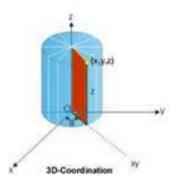
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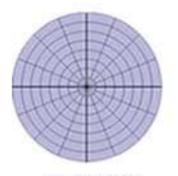


Simulation Methodology

Computational Domain







Top View Grid

- 1. Depth 2900 m
- 2. Radius 3000 m
- 3. Grid Nodes 50-1-10
- 4. Injection Well 0.4 m
- 5. Injection Rate 10 kg/s
- 6. Inj. Temperature 40-70 C
- 7. Inj. Pressure 16-25 MPa





Simulation Methodology

Simulation Approach

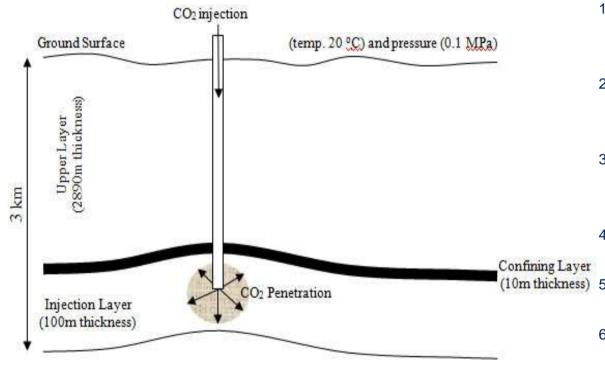
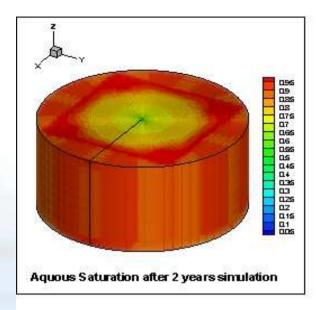


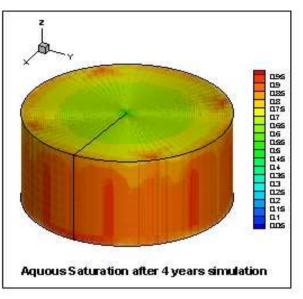
Fig. 3.4 Schematic illustration of CO₂ sequestration in deep saline aquifer.

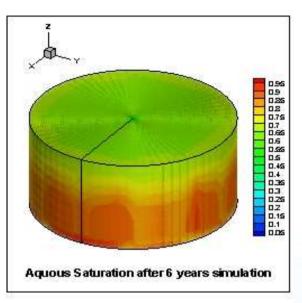
- Quasi-static and dynamic experiments at diff. Time steps.
- 2. Equilibrium reached when no saturation change in all nodes
- 3. Water replacement occur when $P^c = P^d$
- 4. Injection into lower 3 layers
- (10m thickness) 5. Av. Pc and Sw calculted
 - 6. Hydros. cond. 50 C & 100 Bar

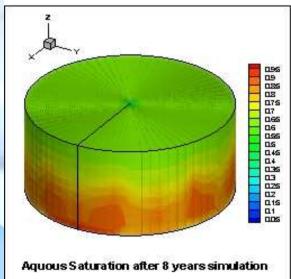


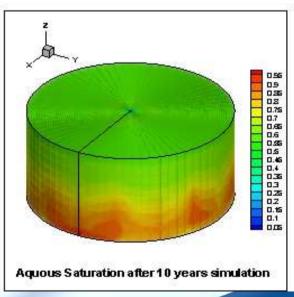
Simulation Results

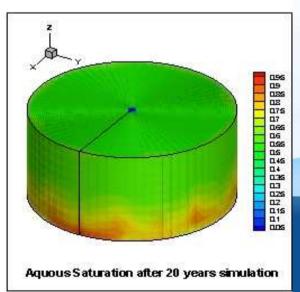






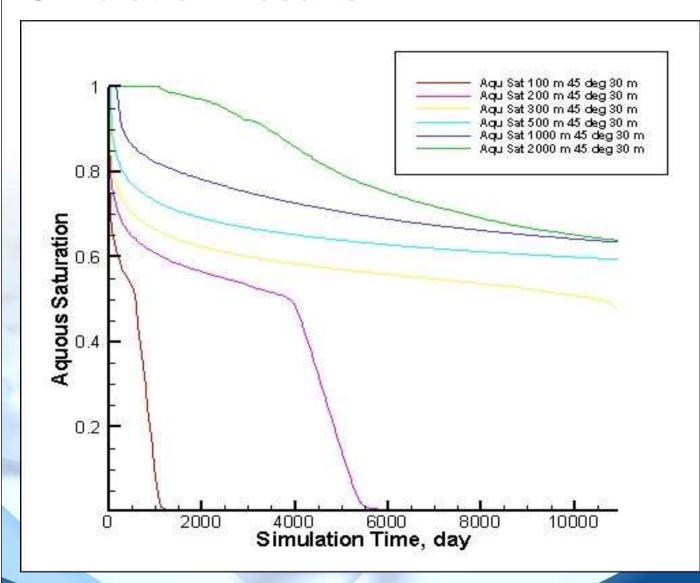








Simulation Results



- Gas Pressure Distribution
- 2. 10 yrs Injection
- 3. 20 yrs Lockup
- 4. Injection into lower 3 layers
- Locally Measured variables



Conclusions

- 1. Deep saline aquifers have a huge feasibility and potential for CO_2 sequestration
- 2. CO₂ storage occurs by structural, residual, solubility and mineral trapping mechanisms
- 3. Buoyancy and capillary forces enhance CO₂ immobilization
- 4. STOMP simulator is utilized to model CO₂ sequestration in saline aquifers for the wide capabilities of analyzing multiple phase flow and all trapping mechanisms
- 5. STOMP post processing data includes using perl script files to convert stomp files to data structured files to be used in Tecplot to plot output files variables.



