# 1 Sensitivity of the early life stages of mayfly to fine sediment and

# 2 orthophosphate levels

- 3 Nicholas C. Everall<sup>1</sup>, Matthew F. Johnson<sup>2</sup>, Paul Wood<sup>3</sup> and Lauren Mattingley<sup>4</sup>
- 4 1. Aquascience Consultancy Limited, Chesterfield, Derbyshire, S42 7JS
  - 2. School of Geography, University of Nottingham, NG7 2RD, UK
    - 3. Department of Geography, Loughborough University, LE11 3TU, UK
    - 4. Salmon and Trout Conservation, Burgate Manor, Fordingbridge, Hampshire, UK

8

9

5

6

7

- Accepted: 31st October 2017
- 10 Published in: Environmental Pollution

11

12

1314

15

16 17

18 19

20

21

22

23

24

25

26

27

28

#### Abstract:

- The ecological effects of interacting stressors within lotic ecosystems have been widely acknowledged. In particular, the ecological effects of elevated fine sediment inputs and phosphate have been identified as key factors influencing faunal community structure and composition. However, while knowledge regarding adult and larval life stage responses to environmental stressors has grown, there has been very limited research on their eggs. In this study, the eggs of the mayfly Serratella ignita (Ephemerellidae: Ephemeroptera) were collected and incubated in laboratory aquaria to hatching under differing concentrations of inert suspended sediment (SS) and orthophosphate (OP), individually and in combination. Results indicate that SS and OP have greater effects on egg hatching in combination than when either were considered in isolation. SS displayed a greater effect on egg survival than OP in isolation or when OP was added to elevated SS treatments. Egg mortality in control treatments was around 6% compared to 45% in treatments with 25 mg l<sup>-1</sup> SS and 52% in 0.3 mg l<sup>-1</sup> OP treatments. Even relatively modest levels of each stressor (10 mg l<sup>-1</sup> SS; 0.1 mg I<sup>-1</sup> OP), below national legal thresholds, had significant effects on egg survival to hatching. The results support calls for legal levels of SS to be reassessed and suggest that more research is required to assess the impacts of pollution on invertebrate egg development given their different sensitivity and exposure pathways compared to other life stages.
- 29 **Capsule:** This study is the first to demonstrate that the survival of mayfly eggs to hatching is 30 significantly reduced by low levels of widespread environmental pollutants in rivers.

#### 1. Introduction

Freshwater organisms are currently subjected by multiple, simultaneous and interacting pressures, due to the co-occurrence of effects associated with climate and land-use change (Tockner et al. 2010; Mantyka-Pringle et al. 2014; Jackson et al. 2016; Sebater et al. 2016). A meta-analysis of research from marine ecosystems has shown that the effect of multiple stressors on aquatic organisms are complex and most frequently synergistic or additive, with the effects being greater or equal when stressors are combined (Przeslawski et al. 2015). In contrast, a recent meta-analysis of freshwater ecosystems reported that the majority of interactions were antagonistic, with effects lower than expected for individual stressors (Jackson et al. 2016); although there has been less research on lotic systems. The uncertainty surrounding the ecological response to multiple, co-occurring stressors can lead to unexpected ecological responses (Christensen et al., 2006; Lindenmayer et al., 2010; Dehedin et al., 2013). For example, Piggott et al. (2012) identified that the negative impacts of fine sediment on invertebrate and algal diversity were greater when water temperature was increased. Holmstrup et al. (2010) reviewed the impacts of multiple stressors on individual organisms, rather than the entire community, and reported that the majority of studies (including temperature, desiccation and chemicals) resulted in synergistic effects.

Aquatic communities are adapted to hydrological regime variability and associated fluxes of solutes and fine sediment (organic and inorganic) derived from the catchment. Lotic ecosystems require sediment inputs to maintain habitat heterogeneity and facilitate nutrient fluxes, but excessive loadings can have negative effects on river ecosystem functioning (Wood and Armitage, 1997; Jones et al. 2012). The US Environmental Protection Agency identified fine sediment deposition as the number one source of stream impairment and habitat degradation nationwide (USEPA, 2000; Evans-White et al. 2013). Fine sediment may degrade aquatic faunal communities and directly affect individual organisms due to burial, scour or abrasion of soft tissues, clogging of respiration structures (gills of invertebrates and fish), as well as reducing habitat quality and increased emigration from degraded habitats (e.g. Billota and Brazier, 2008; Béjar et al. 2017). In addition, fine sediments can reduce habitat availability by covering coarser sediments, filling interstices and modifying biogeochemical conditions by reducing dissolved oxygen concentrations whilst leading to elevation of the concentrations of pollutants within the substrate (Kemp et al., 2011; Jones et al., 2012; Descloux et al. 2014; Mathers et al. 2017). The majority of research centred on fine sediment deposition on aquatic organisms has focused on invertebrate larval community composition or adult life stages (Roy et. al., 2003; Extence et. al., 2013; Bona et al. 2016). For example, the detrimental effects of fine sediment on freshwater mussel population has been examined in detail given that many species are national or internationally endangered and have important functional roles in

rivers (Denic & Geist, 2015; Lummer et al. 2016). However, with the exception of salmonid fish (e.g. Grieg et al. 2005; Jensen et al. 2009; Sternecker & Geist 2010; Chapman et al. 2014), few studies have considered the effects of enhanced fine sediment loading on the egg / embryonic life stages of aquatic fauna.

67

68

69

70

71

72

73

74

75

76

77

78

79

80

81

82

83 84

85

86

87

88

89

90

91

92

93

94

95

96

97

98

99

100

The effects of elevated phosphorus concentrations on aquatic environments, particularly the proliferation of nuisance phytoplankton and both epiphytic and benthic algae has been widely documented (Mainstone and Parr, 2002; Evans-White et al. 2013; Azevedo et al. 2015) and represents a significant threat to water quality and environmental integrity, internationally (Nijboer and Verdonschot, 2004; Smith and Schindler, 2009; Javie et al. 2015). It is well established that nutrient enrichment (eutrophication) has resulted in the reduction of macroinvertebrate community richness through the extirpation of sensitive taxa, particularly within the insect orders Ephemeroptera, Plecoptera and Trichoptera (Ortiz and Puig, 2007; Friberg et. al. 2010; Bini et al. 2014). Orthophosphate (OP) or 'soluble reactive phosphorus' is bioavailable to freshwater organisms and the exceedance of the OP standard has been identified as the single largest cause of water bodies not achieving 'good ecological status' in the UK, under the European Union Water Framework Directive (WFD) (Environment Agency 2012). Phosphorous concentrations have increased in many regions, often linked to human and animal waste; for example, concentrations of Total Dissolved Phosphorous increased by 2000% between 1970 and 2000 in northern Chinese Rivers (Strokal et al. 2016). Phosphorous can be particularly problematic because ecological recovery does not necessarily follow a reduction of concentrations in the environment due to lag times in ecological responses, complex indirect impacts of elevated phosphorous on aquatic communities, and the effects of associated stressors (Javie et al. 2013). In addition, phosphorous can be bound to sediment and remobilised at a later date when phosphorous inputs into the system may be negligible (Meng et al. 2014; Wood et al. 2015; Emelko et al. 2016). Internationally, elevated nutrient and sediment loads are a management priority and are acknowledged to be the primary contributing factor to over 40% of US waters being in poor biological condition (Evans-White et al. 2013).

Some pollutants may have potentially greater effects on early life stages of aquatic biota as they are typically the least mobile and therefore the most vulnerable to disturbance events (Clements and Newman, 2002; Przeslawski et al. 2015). Despite a substantial literature on fish eggs and sedimentation (e.g. see Kemp et al. 2011), relationships between aquatic invertebrates (e.g. Denic and Geist, 2015) and especially their egg survival and environmental stressors are almost completely lacking (but see Gleason et al. 2003; Kefford et al. 2010). Therefore, this study focuses on the effects of increasing suspended sediment (SS) and OP concentrations individually and in combination on the survival and hatching success of the eggs of a widespread and ecologically important aquatic insect

- larvae, *Seratella ignita* (Ephemerellidae: Ephmeroptera) under experimental conditions. This was achieved by investigating whether:
- 103 1. elevated SS concentration impaired egg survival and hatching.
  - 2. elevated OP concentration impaired egg survival and hatching.
- 3. higher concentrations of SS and/or OP had greater effects on egg survival and hatching thanlower concentrations.
  - 4. SS and OP in combination effect hatching / survival to a greater degree than in isolation.

## 2. Methodology:

## 2.1. Target Organism

The Blue-Winged Olive Mayfly (*Serratella ignita* (Poda, 1761): Ephemeroptera: Ephemerellidae) is one of the most common Ephemeroptera species in the British Isles and is present across most of Europe, including the Mediterranean region. Typically nymphs are found in unpolluted, fast flowing systems, emerging between June and September, with nymphs present in the river from March to September (Elliot & Humpesch, 2010; Macadam and Bennett, 2010), although this varies depending on thermal regime and flow permanence (Lopez-Rodriguez et al. 2009). Their life cycle typically includes a long overwintering period in the egg stage. Females of *S. ignita* produce a ball of eggs attached to the posterior underside of the abdomen. The animal descends to the water surface, releasing the egg mass which sinks and becomes anchored to the substrate via fibrous attachments (Gaino & Bongiovanni, 1992). *S. ignita* is ecologically important because of its widespread distribution and high abundance, which makes it significant for supporting fisheries. However, numbers have declined in a number of UK rivers over the past 20 years, particularly chalk streams (Bennett & Gilchrist, 2010). *S. ignita* larvae are known to be sensitive to fine sediment loading and OP concentration, with investigations linking losses of *S. ignita* to enhanced fine sediment loading effects in European rivers (Everall, 2010; Larsen *et. al.*, 2011; Minutoli *et al.* 2013).

## 2.2. Experimental Set-Up and Overview

Experiments were undertaken in experimental, laboratory chambers (Figure 1). A total of 24 chambers were run in parallel for the duration of the experimental period, each representing a different treatment. Each experimental chamber housed 3 glass laboratory slides, which acted as a substrate for *S. ignita* eggs. Each slide contained approximately 30 egg masses which were left to

develop on the slides for 8 months under either control or experimental treatment conditions. The experimental chambers consisted of plastic funnels with a slide mount lodged above the outflow. The 3 glass slides were held vertically at 45 degrees and in parallel in the slide mount and remained submerged for the duration of the experiment in the 20 mm diameter circular container.

The water used in experiments was aerated and had pre-determined concentrations of OP and SS. It was held in 25 I reservoirs, elevated above the experimental chamber and allowed to flow through the system under gravity from a tap, which limited the flow rate to 0.65 ml min<sup>-1</sup>. Water drained through the experimental chamber and out through a pipe to a drain (Figure 1).

# 2.3. Serratella ignita egg collection and laboratory acclimation

Hundreds of swarming gravid adult *Serratella ignita* were collected from above the water surface of the River Manifold in Staffordshire, UK (53°09′49.15″N; 001°51′35.70″W) in August 2015. Adult *S. ignita* were carefully transferred in ventilated plastic aquaria (40 x 25 cm) fitted with temporary cardboard floors to the laboratory where they were placed on top of white plastic trays (38 x 22 x 5 cm). Each tray bottom was lined with 36 sterilised glass slides and covered by 2 cm of aerated water from the sample site. The temporary cardboard floors of the adult mayfly aquaria were removed allowing gravid female *S. ignita* to access the river water surface in the glass slide lined trays to lay their eggs.

Over 24 hours, the gravid female *S. ignita* laid their egg masses on to the water surface whereupon they sank and became attached to the glass slides lining the trays. Spent *S. ignita* spinners and body parts (Supplementary Material A) were carefully removed from the egg slides using sterilised steel forceps, paying attention not to disturb the deposited egg masses. The egg mass covered slides were left *in situ* for another 24 hours to allow egg mass adhesion to the glass slides after which they were transferred to slide holders in the treatment chambers (Supplementary Material B) using sterilised forceps. Prior to transfer, each slide had the number of egg masses per slide recorded in indelible ink on the slide. All 24 experimental chambers had been running for 20 days with a discharge flow of 0.65 ml min<sup>-1</sup> of carbon-limestone filtered tap water prior to slide introduction.

All of the egg mass slides were left to acclimate in flow through chambers supplied with dechlorinated, filtered tap water for a further 24 hours prior to the commencement of experiments with treatment exposures. The acclimation and treatment bioassays were subject to ambient outdoor air temperatures, humidity and light regime during the experimental period between August 2015 and March 2016.

## 2.4. Bioassay design and egg monitoring

A summary of the bioassay treatments used in the experiment are presented in Table 1. Every month of the bioassay the 3 slides in each treatment were placed under a microscope for a few minutes in a wet mount containing the appropriate bioassay test solution for observation. The slides were examined for any egg mass loss, egg mass emergence, and secondary biological growth e.g. fungal hyphomycae. Egg mass emergence was considered when approximately >90% of the viable eggs in the mass had hatched. Egg mass loss was considered to have occurred if a similar proportion of the individual eggs within the mass had died or if the egg mass had fallen from the slide, identification of which was aided by marks left on the slide by displaced egg masses (Supplementary material C). Secondary fungal hyphomycae growth was clearly identifiable under a microscope. In subsequent analysis, the status of egg masses was aggregated across the 3 slides.

After 3 months the slides in each treatment were carefully checked monthly for individual egg mortalities, within egg masses. It was not possible to count every egg within all egg masses. Instead, an egg mass was randomly selected from each of the 3 slides in each experimental chamber and the state of 200 eggs were counted under a microscope. Eggs were recorded as being either healthy or dead, where dead eggs were readily identified because they turned an opaque white, as reported previously by Yeo and Dechoretz (1973). Any egg mass chosen for egg mortality observation had an indelible dot placed on the reverse side of the slide so that it was not chosen again for observation. In subsequent analysis, egg mortality was aggregated across the 3 slides in each experimental chamber, giving a total sample of 600 individual eggs for each treatment.

#### 2.5. Chemical dosing and testing

The base and experimental control water during the acclimation and bioassay testing period was tap water run at 0.65~ml min<sup>-1</sup> through filters housing a mix ratio of 5:1~6~mm limestone chippings: granular activated carbon to reduce background levels of OP in tap water and remove any trace impurities. Test compounds were made up from the base water with the required dose additions of Sigma Aldrich 1000 mg l<sup>-1</sup> Orthophosphate and 1000 mg l<sup>-1</sup> Total Suspended Solids (inert silica particles, diameter  $5-100~\mu\text{m}$ ) calibration standards.

Water temperature in experimental test chambers was recorded daily in the control and once a month in other treatments. Daily water temperature mirrored ambient air temperature

(Supplementary material D). Dissolved oxygen and pH were also recorded once a month across all the treatments. Water samples were taken from all of the bioassay test chambers once a month across the 8 month experimental period using an overflow valve fitted into the treatment rigs (Figure 1). All samples for chemical analysis were sent to the UKAS Accredited National Laboratory Service for monthly analyses of Total Nitrogen, Ammoniacal Nitrogen, Nitrite, Alkalinity (to pH 4.5 as CaCO3), Orthophosphate, pH, Suspended Solids (at 105 °C), Boron, Calcium, Iron, Lithium, Magnesium, Manganese, Sodium, Water hardness (Total as CaCO3), Arsenic, Selenium, Cadmium, Copper, Lead, Mercury, Nickel and Zinc (Supplementary material E). The test treatments were also tested monthly for respective actual dosed SS and OP levels.

Mean physical-chemical properties for each of the bioassay chambers from 8 monthly samples across the study period are presented in Table 2. Individual monthly measurements of physical-chemical conditions are presented in Supplementary material D and clearly indicate that background water quality was stable with low level of trace chemicals in the control and test diluent water during the tests (Supplementary material E). With the exception of the test variables, there were no significant differences found between diluent physical-chemical conditions in the bioassays across the 8 month experiment (p < 0.01; ANOVA). The measured concentrations of SS and OP in water samples also display good spatial and temporal stability with the dosed concentration. In control experiments, OP concentrations were ~0.04 mg l<sup>-1</sup> despite ~96-98% phosphate removal from the baseline diluent tap water, similar to the performance of phosphorous removal of other workers using this type of filter (e.g. Hussain *et. al.*, 2011).

# 2.6. Statistical analysis

A regression modelling approach was used to examine the impact of concentration gradients of OP and SS on egg mortality. The number of dead eggs recorded over time, from a sub-sample of 600 eggs, was recorded for different concentrations and combinations of OP or SS. Regression models were developed for the association between the total number of dead eggs and the concentration of OP and/or SS for different exposure periods. In addition, regression models were developed for the association between the percentage of egg masses that emerged at the end of the experiment and OP and SS concentrations, and combinations to examine additive effects.

#### 3. Results:

## 3.1. Egg mortality within bioassays

The number of dead eggs recorded within egg masses increased as the concentration of OP and SS increased above control levels. Egg mortality in control experiments were consistent between treatments and remained low, averaging 5.8% of sampled eggs across all treatments and ranging from 27 to 42 eggs out of 600. As concentration of SS and OP increased, there were substantial increases in egg mortality, representing a 972% increase under the highest OP levels and 1261% increase under the highest SS concentration over control levels. Mortality increased exponentially with SS and OP concentration when dosed individually (Figure 2a; b), with significant regression models developed between OP or SS concentration and mortality after 71 days of exposure (SS p < 0.01,  $R^2 = 0.88$ ; OP p < 0.01,  $R^2 = 0.99$ ). After 183 days of exposure, exponential relationships between OP or SS concentration remained significant (p < 0.01 in both cases) with high explanatory power (98% of variance in both cases) (Table 3).

When SS and OP were dosed in combination, mortality increased over equivalent concentrations in isolation (Figure 2c; d). The increase in egg mortality when 0.07 mg l<sup>-1</sup> OP was added was small, but consistent and the relationship remained exponential. In contrast, the addition of 10 mg l<sup>-1</sup> of SS to OP concentrations resulted in a marked increase in mortality and a change in the relationship between egg mortality and OP concentration from exponential to linear (Table 3).

In control runs and low doses of SS and OP (<5 mg  $l^{-1}$  and < 0.1 mg  $l^{-1}$ , respectively), egg mortality did not increase over time, but remained around 6% of sampled eggs (Figure 3a; b). Although mortality increased when SS was elevated to above 10 mg  $l^{-1}$ , egg mortality did not increase substantially over time, only increasing by about 10% of the sampled eggs over the duration of experiments (10 mg  $l^{-1}$  10% to 20%; 15 mg  $l^{-1}$  19% to 29%). When SS was above 20 mg  $l^{-1}$ , mortality increased through time linearly, from 45% to 80% of sampled eggs in the case of the highest dose (Figure 3a). For OP, mortality consistently increased through time for all treatments except the control; however this was limited to less than 6% for all treatments, except the highest two concentrations (Figure 3b). Similar patterns were observed when OP and SS were dosed in combination, with egg mortality increasing through time with the rate of mortality increasing as dosage increased. When SS was added to OP treatments, egg mortality increased faster and to a higher percentage of sampled eggs in comparison to when OP was added to SS treatments (Figure 3c, d).

## 3.2. Egg mass emergence in bioassays

The number of egg masses that emerged decreased exponentially as concentration of OP or SS increased (OP  $R^2 = 0.998$ ; SS  $R^2 = 0.963$ ; p < 0.01 in both cases). OP effects were discernible from control treatments at 0.1 mg  $I^{-1}$  and from SS controls at 10 mg  $I^{-1}$  (Figure 4a, b). The emergence of egg masses exposed to OP declined substantially when 10 mg  $I^{-1}$  of suspended sediment was added to treatments, supporting the findings of individual egg counts (Figure 4a). An exponential relationship between egg emergence and OP persisted with the addition of SS (p < 0.01;  $R^2 = 0.998$ ) but with greater egg mass emergence at concentrations of OP 0.1 mg  $I^{-1}$  and above. In contrast, when 0.07 mg  $I^{-1}$  of OP was added to SS treatments, there was no clear difference between egg mass emergence with and without OP (Figure 4b).

The 3 separate slides within each treatment indicated very high consistency in results observed. The largest difference in egg mass emergence between the 3 slides within each treatment was 27% for those subjected to 0.3 mg l<sup>-1</sup> OP plus 10 mg l<sup>-1</sup> SS (Figure 5).

## 4. Discussion:

## 4.1. Effects of multiple stressors

When low levels of SS were added to OP treatments, the mortality rate of eggs increased markedly, indicating that SS and OP had a greater impact on *S. ignita* when combined than when present individually. However, when low levels of OP were added to SS treatments, there was no discernible effect on mortality rates above SS in isolation, suggesting SS had a greater effect on egg and egg mass survival than OP.

Studies focusing on multiple stressors have consistently reported fine sediment to be a more pervasive stressor to the abundance of individual invertebrate species (Wagenhoff *et. al.*, 2011) and invertebrate communities (Piggot et al. 2015; ; Elbrecht et al. 2016) than enhanced nutrient concentrations, indicating that priority should be given to minimising fine sediment over nutrient inputs. However, contrasting results have been found in some cases where chemical composition of fine sediment was more important than sediment quantity in controlling invertebrate community composition (von Bertrub et al., 2013). For example, Andersen et al. (2006) found uncontaminated sediments had no effect on the survival of several invertebrate species (*Hyalella Azteca* [amphipoda; Hyalellidae]; *Procloeon* sp. [Ephemeroptera; Baetidae]; *Chironomus dilutes* [Diptera; Chironomidae]). In the current experiments, suspended sediment was inert silica particles, clearly demonstrating that it is the deposition of sediment, rather than associated chemicals, that effected *S. ignita* egg

development. A potential explanation for the difference between studies is that the species examined by Andersen et al. (2006) were characteristic of slow flowing lowland streams or marginal habitats dominated by macrophytes, where fine sediment concentrations and accumulations were naturally high. This contrasts with *S. ignita*, which is typical of moderate flowing streams with coarse substrates and where SS concentration is likely to be much lower than lowland reaches. As such, *S. ignita* is adapted to environments with naturally lower concentrations of fines and as a result fine sediment potentially acts as a stressor at lower concentrations than for many species adapted to slow flowing habitats (Elliot and Humpesch, 2010). Consequently, it is likely that the relative significance of SS and associated contaminates will depend on the receptor species and their association with specific habitats.

Each female *S. ignita* produces many eggs and as a result the effect of the elevated egg mortality on the viability of populations is difficult to assess. For example, it is possible that if hatching success was high in rivers, density dependent processes may result in many early instar larvae perishing, reducing the population level effects of egg mortality due to anthropogenic stressors. Therefore the results here do not necessary imply population level impacts. In addition, *S. ignita* is a common species in the UK and Europe and often occurs in high abundance. However, *S. ignita* larvae have been shown to be highly sensitive to sedimentation and their presence is used in national biological metrics to indicate reduced fine sediment pressure (Extence et al. 2011). In addition, the proportion of individuals commonly surviving through to reproduction is not known and there is both anecdotal and documented evidence that the abundance of *S. ignita* has declined over the past 20 years in some English rivers (Bennett & Gilchrist, 2010).

# 4.2. The effect of suspended sediment on eggs

The results of this study indicate that the egg stage of *Serratella ignita* is susceptible to sustained high levels of SS concentration during their 8 month developmental period. Concentration dependent mortality of *S. ignita* eggs was evident at annual mean equivalent concentrations of 10 - 25 mg l<sup>-1</sup> but levels of fine sediment < 10 mg l<sup>-1</sup> displayed no markedly higher egg mortality than the control treatments. The cause of egg mortality is hypothesised to be reduced oxygen transfer due to sediment coating egg surfaces. Additionally, the build-up of fine sediment over time caused some of the egg masses to be dislodged from the slides after 6 months of exposure in the 20 and 25 mg l<sup>-1</sup> treatments. These egg masses were eroded and lost within the dosing rig sumps and, therefore, it is

not clear if the individual eggs were still viable; however, in watercourses dislodgement exposing egg masses to scour damage, burial and predation would be highly disadvantageous.

Other experiments examining the effect of fine sediment covering on invertebrate eggs have reported reduced survival and hatching for *Chironomus cloacalis* (Diptera; Chironomidae), *Physa acuta* (Gastropod; Physidae) and *Gyraulus tasmanica* (Gastropod; Planorbidae) (Kefford et al. (2010). In control treatments without fine sediment, 100% of viable eggs of all three species hatched, but this was reduced when buried with clay (kaolin) or sand; although, the direct effects of suspended sediment were limited. Similarly, Gleason et al. (2003) found that burial to 0.5 cm caused a 99.7% reduction in the emergence of invertebrate eggs from wetlands. The impact of SS on invertebrates is complex because of associated contaminates; for example, the source of sediment has been shown to be important for salmonid embryo development, primarily because the organic matter content of sediment consumes oxygen as it degrades, potentially reducing oxygen availability to developing embryos (Sear et al. 2014). In addition, the influence of other stressors confound ecological response; for example, Doretto et al. (2017) found that high availability of coarse particulate organic matter mitigated the negative effects of fine sediment, which clogged interstitial spaces in artificial substrates in the Po River, Italy.

Fish eggs are negatively effected by fine sediment (Kemp et al. 2011). Much of the research on fish embryo development and fine sediment has focused on the clogging of interstitial spaces in salmonid fish redds and associated reduction of interstitial flow volume and velocity (Jensen et al. 2009; Chapman et al. 2014). However, deposition of clay particles directly onto salmon (*Salma salar*) eggs has been shown to reduce oxygen exchange across the egg membrane and increase mortality (Greig et al. 2005). This mechanism is also hypothesised to be responsible for mayfly egg mortality in these experiments. Research has also demonstrate that salmonid fish egg development can be effected by sedimentation by the prevention of the expulsion of metabolic wastes from the egg chorion (Chapman 1988; Bennett et al. 2003). Concentrations of nitrates and ammonia may significantly affect salmonid egg development (e.g. Sternecker et al. 2013) and Reynolds & Guillaume (1998) found phosphate concentrations of 0.5 mg l<sup>-1</sup> resulted in earlier emergence of European Bitterling (*Rhodeus sericeus*) embryos from eggs deposited within the gills of freshwater mussels. However, little research has investigated the link between elevated phosphorous concentration and fish embryo development.

## 4.3. Phosphorous effects on egg development

Mortality of *S. ignita* eggs was evident at annual mean equivalent concentrations of  $0.1 - 0.3 \text{ mg l}^{-1}$  OP but levels of biologically available phosphorous <  $0.1 \text{ mg l}^{-1}$  resulted in no higher egg mortality than in control treatments. The cause of egg mortality in the highest dose of  $0.3 \text{ mg l}^{-1}$  appeared to be related to the growth of aquatic fungal filaments smothering the egg masses after 1 month of exposure. The adhesive, mucous coating of mayfly eggs has been postulated to protect the egg from bacterial and fungal attack (Gaino *et. al.*, 2009), although any protection appeared to have been lost as a result of elevated OP stimulating microbial growth, resulting from the high availability of phosphorous. From light microscopy examination of fungal smothered eggs both undetermined aquatic hyphomycete species and *Fusarium aquaeductum* were identified coating the egg surfaces. Aquatic hyphomycetes growing on fish eggs have been found to be pathogenic (Wedekend *et. al.*, 2010) and *Fusarium* species have been documented parasitizing eggs of the Penaeid prawn *Marsupenaeus japonicus* (Momoyama, 1987).

Egg mortality in treatments with lower OP levels, where there was no evidence of fungal growth, suggested other direct impacts of elevated biologically available phosphorous levels. Chronic exposure to sub-lethal concentrations of phosphates have been reported to have negative effects on early stages of aquatic fauna; for example, abnormal embryonic development in sea urchin (*Lytechinus variegatus*) (Bottger and McClintock, 2002). In addition, the cells of more complex organisms have also shown impaired gene expression (Rutherford *et. al.*, 2006) and cell membrane scrambling (Voelkl *et. al.*, 2014) with increasing extracellular phosphate concentrations. Therefore, it is hypothesised that egg mortality in the range of continuous OP exposures of 0.1 – 0.2 mg l<sup>-1</sup> may have been due to direct physiological and genotoxic impacts.

# 4.4. Concentrations of OP and SS in rivers

The ability of an organism to survive exposure to a stressor is dependent upon the concentration of the parameter and duration of exposure (Tabak and Gibbs, 1991; Zhao and Newman, 2006; Cope et al. 2008). The total duration of exposure to a concentration of suspended solids is acknowledged to be a key variable determining its effect on aquatic biota (Billota and Brazier, 2008). For example, Maturana *et al.* (2014) found that continuous, chronic exposure of sediment had a greater detrimental impact on salmonid embryos than instantaneous pulses of sediment. The exposure conditions here were not directly comparable to natural conditions within a river, where pulsed and intermittent exposure of organisms to sediments and nutrients are common (Alabaster and Lloyd, 1980; Davies and Bothwell, 2012; Outram et al. 2014). However, SS and OP levels used in these

experiments represent relatively modest concentrations for many English rivers, typically below WFD specified thresholds.

384

385

386

387

388

389

390

391

392

393

394

395

396

397

398

399

400

401

402

403

404

405

406

407

408

409

410

411

412

413

414

415416

The Environment Agency (EA), the statutory environmental regulator in England, recorded 32549 spot measurements of OP across England in 2015 and 22% of those were above 0.3 mg l<sup>-1</sup>, the highest concentration used in these experiments. Monthly spot measures were made at 1812 locations across England in 2015, where at least 9 measurements were made throughout the year. Of these sites, 22% had annual average values higher than 0.3 mg l<sup>-1</sup> and over half had average concentrations above 0.1 mg l<sup>-1</sup>, the lowest concentration used in these experiments with a statistical effect on egg survival. 26% of sites had OP levels above 0.1 mg l<sup>-1</sup> in every measurement made throughout the year (Table 4). These results are consistent with the work of Worrell et al. (2016) who calculated that annual average OP concentrations have declined from 0.19 to 0.1 mg | 1 between 1974 and 2012 in England, Wales and Scotland, based on routine monitoring data. Therefore, whilst concentrations of OP are declining across Europe (Bouraoui and Grizzetti, 2011), in many streams concentrations remain above those found to exert an effect in the experiments reported in this study. Fewer sites were sampled for suspended sediment but of the 129 sites with 9 or more measurements in 2015, 9% had average values higher than 25 mg l<sup>-1</sup> and 39% had values over 10 mg l<sup>-1</sup>, found to effect egg hatching in these experiments (Table 4). OP legal levels are dependent on site specific characteristics and physio-chemical conditions but it is clear that low-tomoderately elevated levels can have direct effects on insect egg development, which may be accelerated at higher concentrations by fungal growth. In addition, elevated OP in rivers alters primary production leading to important indirect implications for dissolved oxygen concentrations and water temperature because excessive plant and algae growth can shade the water column, which may also impact insect egg development (Humpesch 1980; Elliot, 1987; Pritchard et al. 1996; Bennett 2007; Rotvit and Jacobsen, 2013).

The current experimental findings support the growing concern that the annual mean SS guideline standard of 25 mg  $I^{-1}$  in the UK is not sufficient (WWF, 2007). This is supported by other studies that have identified effects of fine sediment on invertebrate survival at levels  $\geq 8$  mg  $I^{-1}$  in Canadian freshwaters (Rosenberg and Wiens, 1978; Quinn *et. al.*, 1982). In these experiments, egg masses were lost because of sediment coverage and the weight of deposited sediment dislodging them, although it is not clear whether dislodgement occurred during or after egg health had deteriorated, potentially reducing their adhesive properties. In rivers, this would probably result in the burial and/or damage of eggs. The coating of eggs with sediment has implications for oxygen transfer which will be partly controlled by the extent of sediment coverage on the egg surface, as well as the

particle size and shape. These parameters will be at least partially dependent on the flow velocity and sediment properties and are likely to be less well correlated to suspended sediment concentrations. Therefore, the results support the assertion of Bilotta and Brazier (2008) and Kefford *et al.* (2010) that standards should move away from turbidity or suspended sediment concentrations to focus on settlement rates and sediment properties. Similarly, the source of sediment could have different effects on egg mortality because of its ability to harbour other pollutants, including phosphorous.

424

425

426

427

428

429

430

431

432

433

434

435436

437

438

439

440

441

442

417

418

419

420

421

422

423

#### 4.5. Management implications

Previous research on the larval and adult stage of invertebrates indicates that elevated SS and OP are pervasive issues in river management (Friberg et. al. 2010; Jones et al. 2012; Bini et al. 2014; Mathers et al. 2017). Internationally, OP concentrations remain high and rising in many river systems, in particular due to agricultural intensification and population increases coupled with the direct discharge of untreated human waste (Tysman et al. 2013; Strokal et al. 2016; Yan et al. 2016). Despite reductions in both SS and OP concentrations in river systems across Europe and North America, many rivers still show clear signs of negative impact (Javie et al. 2015; Blaas and Kroeze, 2016). This is likely to be partly related to the indirect impacts of OP and SS, their interaction with other stressors, lags in ecological response, and remobilisation of OP bound to sediments long after inputs into the river system have been reduced (Jarvie et al. 2012). However, the results presented here suggest that relatively low levels of both SS and OP can negatively affect invertebrate egg development. Therefore, it is possible that by focusing on the larval and adult stage of invertebrate development, important information is being missed about the tolerance of species during what is potentially their most vulnerable developmental stage. More information is needed on the effect of stressors on egg development as impaired hatching could have significant implications for invertebrate populations at lower pollutant concentrations than those observed to effect larval and adult stages of the same species.

443

444

445

446447

#### 5. Conclusions:

The effects of environmental pollutants on the eggs of aquatic invertebrates are not well understood despite the fact that eggs are potentially the most vulnerable life stage of many invertebrates. Relatively modest levels of SS and OP have highly significant detrimental effects on the mortality of

*S. ignita* eggs, with potentially significant implications for populations of mayfly. Fine sediment was the more pervasive stressor, increasing mortality of eggs exposed to OP enrichment, whereas elevated OP levels did not significantly increase mortality in comparison to those exposed only to fine sediment. The direct mechanism for the detrimental effects on eggs is likely to be complex but suspended sediment settled onto eggs, coating them and under high dosage (> 0.2 mg  $I^{-1}$ ) resulting in dislodgement. High OP levels (> 0.2 mg  $I^{-1}$ ) fuelled the growth of hyphomycete, which negatively affected eggs. The mechanism by which lower levels of OP (0.1 – 0.2 mg  $I^{-1}$ ) negatively impacted eggs, in the absence of hyphomycete growth, is not known. Current legal limits of SS and OP in the European Union are above those found to have an effect in the experiments reported in the study and suggests management needs to focus on elevated SS levels. Although levels are dropping across Europe – substantially in the case of OP – the results of these experiments support growing concern about current guidelines relating to SS and associated organic contaminants and the need for more stringent regulation.

# Acknowledgements

We would like to thank Salmon and Trout Conservation UK for funding the chemical analysis and Ben Heynes for assistance with field work.

- 466 **References**:
- 467 Alabaster, J.S. and Lloyd, R. (1980) Water quality criteria for freshwater fish. Food and agriculture
- organization of the United Nations, Butterworths, London.
- 469 Anderson, B.S., Phillips, B.M., Hunt, J.W., Connor, V., Richard, N. and Tjeerdema, R.S. (2006)
- 470 Identifying primary stressors impacting macroinvertebrates in the Salinas River (California, USA):
- 471 relative effects of pesticides and suspended particles. *Environmental Pollution* **141**: 402–408.
- 472 Azevedo, L.B., van Zelm, R., Leuven, R.S.E.W., Hendriks, A.J. and Huijbregts, M.A.J. (2015) Combined
- 473 ecological risks of nitrogen and phosphorus in European freshwaters. *Environmental Pollution*
- **200**: 85–92.
- 475 Béjar, M., Gibbins, C.N., Vericat, D. and Batalla, R.J. (2017) Effects of suspended sediment transport
- on invertebrate drift. *River Research and Applications*.
- Bennett, C.J. (2007) The ecology of a seven year study of the life cycle of the mayfly *Ephemera*
- 478 danica. Freshwater Forum **27**: 3–14.
- 479 Bennett, C and Gilchrist, W. (2010). Chapter 22. Riverflies. In: Maclean, N. (Ed) Silent Summer. The
- 480 State of Wildlife in Britain and Ireland. Cambridge University Press, Cambridge, UK.
- 481 Bennett, D.H., Connor, W.P. and Eaton, C.A. (2003) Substrate composition and emergence success of
- fall Chinook Salmon in the Snake River. *Northwest Science* **77**: 93 99.
- 483 Billota, G.S. and Brazier, R.E. (2008). Understanding the influence of suspended solids on water
- quality and aquatic biota. Water Research 42: 2849-2861.
- Bini, L.M., Landeiro, V.L., Padial, A.A., Siqueira, T. and Heino, J. (2014) Nutrient enrichment is related
- to two facets of beta diversity for stream invertebrates across the United States. *Ecology* **95**:
- 487 1569–1578.
- 488 Blaas, H. and Kroeze, C. (2016) Excessive nitrogen and phosphorus in European rivers: 2000 2050.
- 489 Ecological Indicators 67: 328 337.
- 490 Bona, F., Doretto, A., Falasco, E., La Morgia, V., Piano, E., Ajassa, R. and Fengoglio, S. (2016)
- 491 Increased sediment loads in alpine streams: an integrated field study. River Research and
- 492 *Applications* **32**: 1316–1326.

493 Bottger, S.A. and McClintock, J.B (2002) The effects of organic and inorganic phosphates on 494 fertilization and early development in the sea urchin Lytechinus variegatus (Echinodermata: 495 Echinoidea). *Journal of Experimental Zoology* **292**: 660–671. 496 Bouraoui, F. and Grizzetti, B. (2011) Long term change of nutrient concentrations of rivers 497 discharging in European seas. Science of the Total Environment 409: 4899–4916. 498 Chapman, D.W. (1988) Critical review of variables used to define effects of fines in redds of large 499 salmonids. Transactions of the American Fisheries Society 117: 1-21. 500 Chapman, J.M., Proulx, C.L., Veilleux, M.A.N., Levert, C., Bliss, S., André, M-È., Lapointe, N.W.R. and 501 Cooke, S.J. (2014) Clear as mud: A meta-analysis on the effects of sedimentation on freshwater 502 fish and the effectiveness of sediment-control measures. Water Research 56: 190–202. 503 Christensen, M.R., Graham, M.D., Vinebrooke, R.D., Findlay, D.L., Paterson, M.J. and Turner, M.A. 504 (2006) Multiple anthropogenic stressors cause ecological surprises in boreal lakes. Global Change 505 *Biology* **12**: 2316 – 2322. 506 Clements, W.H. and Newman, M.C. (2002) Community Ecotoxicology. Hierarchical Ecotoxicology 507 Series. John Wiley and Sons Ltd, Hoboken, USA. 508 Cope, W.G., Bringolf, R.B., Buchwalter, D.B., Newton, T.J., Ingersoll, C.G., Wang, N., Augspurger, T., 509 Dwyer, F.J., Barnhart, M.C., Neves, R.J. and Hammer, E. (2008) Differential exposure, duration, 510 and sensitivity of unionoidean bivalve life stages to environmental contaminants. Journal of the 511 North American Benthological Society 27: 451–462. 512 Davies, J.M. and Bothwell, M.L. (2012) Responses of lotic periphyton to pulses of phosphorus: P-flux 513 controlled growth rate. *Freshwater Biology* **57**: 2602–2612. 514 Dehedin, A., Maazouzi, C., Puijalon, S., Marmonier, P. and Piscart, C. (2013) The combined effects of 515 water level reduction and an increase in ammonia concentrations on organic matter processing 516 by key freshwater shredders in alluvial wetlands. Global Change Biology 19: 763-774. Denic, M. and Geist, J. (2015) Linking stream sediment deposition and aquatic habitat quality in Pearl 517 518 Mussel streams: Implications for conservation. River Research and Applications 31: 943–952. 519 Descloux, S., Datry, T. and Marmonier, P. (2014) Trait-based structure of invertebrates along a 520 gradient of sediment colmation: benthos versus hyporheos responses. Science of the Total

521

Environment **466**: 265–276.

- Doretto, A., Bona, F., Piano, E., Zanin, I., Eandi, A.C. and Fenoglio, S. (2017) Trophic availability
- 523 buffers the detrimental effects of clogging in an alpine stream. Science of the Total Environment
- **592**: 503–511.
- 525 Elbrecht, V., Beerman, A.J., Goessler, G., Neumann, J., Tollrian, R., Wagner, R., Wlecklik, A., Piggott,
- J.J., Matthaei, C.D. and Leese, F. (2016) Multiple-stressor effects on stream invertebrates: a
- 527 mesocosm experiment manipulating nutrients, fine sediment and flow velocity. Freshwaer
- 528 *Biology* **61**: 362–375.
- 529 Elliott, J.M. and Humpesch, U.H. (2010) Mayfly larvae (Ephemeroptera) of Britain and Ireland: keys
- 530 and a review of their biology. Scientific Publications of the Freshwater Biological Association 66,
- 531 Elliot, J.M. (1987) Temperature-induced changes in the lifecycle of Leuctra nigra (Plecoptera:
- 532 Leuctridae) from a Lake District stream. Freshwater Biology 18: 177–184.
- 533 Emelko, M.B., Stone, M., Silins, E., Allin, D., Collins, A.L., Williams, C.H.S., Martens, A.M. and Bladon,
- 534 K.D. (2016) Sediment-phosporus dynamics can shift aquatic ecology and cause downstream
- legacy effects after wildfire in large river systems. *Global Change Biology* **22**: 1168–1184.
- 536 Environment Agency (2012) Freshwater eutrophication: A nationally significant water management
- issue. Briefing note for 10.12.2012 workshop. Accessible at: http://www.rivermease.co.uk/wp-
- content/uploads/2015/07/Freshwater\_Eutrophication\_Briefing\_Note\_December\_20121.pdf
- 539 Evans-White, M.A., Haggard, B.E. and Scott, J.T. (2013) A review of stream nutrient criteria
- development in the United States. Journal of Environmental Quality 42: 1002–1014.
- Everall, N.C. (2010) The aquatic ecological status of the rivers of the Upper Dove Catchment in 2009.
- Natural England Commissioned Report NECR046. Natural England: Sheffield.
- 543 Extence, C.A., Chadd, R.P., England, J., Dunbar, M.J., Wood, P.J. and Taylor, E.D. (2013) The
- Assessment of Fine Sediment Accumulation in Rivers Using Macro-invertebrate Community
- Response. River Research and Applications 29: 17 55.
- 546 Friberg, N., Skriver, J., Larsen, S.E., Pedersen, M.L. and Buffagni, A. (2010) Stream macroinvertebrate
- occurrence along gradients in organic pollution and eutrophication. Freshwater Biology 55: 1405–
- 548 1419.
- 549 Gaino, E., Sartori, M. and Rebora, M. (2009) Chorionic fine structure of eggs from some species of
- 550 Proboscidoplocia (Ephemeroptera; Ephemeroidea). Italian Journal Of Zoology 68: 1-8.

551 Gaino, E. and Bongiovanni, E. (1992) The egg of Pseudocentroptilum pennulatum: a scanning 552 electron microscopy (SEM) study (Ephemeroptera, Baetidae). Bolletino della Società 553 Entomoogical Italiana 12: 195–200. 554 Gleason, R.A., Euliss Jr, N.H., Hubbard, D.E. and Duffy W.G. (2003) Effects of sediment load on 555 emergence of aquatic invertebrates and plants from wetland soil egg and seed banks. Wetlands 556 23: 26-34. 557 Grieg, S.M., Sear, D.A. and Carling, P.A. (2005) The impact of fine sediment accumulation on the 558 survival of incubating salmon progeny: Implications for sediment management. Science of the 559 Total Environment 344: 241-258. 560 Holmstrup, M., Bindesbøl, A.-M., Oostingh, G.J., Duschl, A., Scheil, V., Køhler, H.-R., Loureiro, S., Soares, A.M.V.M., Ferreira, A.L.G., Kienle, C., Gerhardt, A., Laskowski, R., Kramarz, P.E., Bayley, 561 M., Svendsen, C. and Spurgeon, D.J. (2010) Interactions between effects of environmental 562 563 chemicals and natural stressors: A review. Science of the Total Environment 408: 3746 – 3762. Humpesch, U.H. (1980). Effect of temperature on the hatching time of eggs of five Ecdyonurus spp. 564 565 (Ephemeroptera) from Austrian streams and English streams, rivers & lakes. Journal of Animal 566 Ecology **49:** 317-333 567 Hussain, M., Hussain, S.M., Afzal, M., Javid, A., Abdullah, S., Raza, S.A. and Iqbal, M. (2011) Effect of 568 low level of mineral phosphorus and medium level of nitrogen on planktonic productivity and increase in fish production. Journal of Animal and Plant Science 21: 642-645. 569 570 Jackson, M.C., Loewen, C.J.G., Vinebrooke, R.D. and Chimimba, C.T. (2016) Net effects of multiple 571 stressors in freshwater ecosystems: a meta-analysis. Global Change Biology 22: 180-189. 572 Javie, H.P., Sharpley, A.N., Flaten, D., Kleinman, P.J.A., Jenkins, A. and Simmons, T. (2015) The pivotal 573 role of phosphorus in a resilient water-energy-food security nexus. Journal of Environmental 574 Quality 44: 1049-1062. 575 Javie, H.P., Sharpley, A.N., Scott, J.T., Haggard, B.E., Bowes, M.J. and Lesley, B.M. (2012) Within-river 576 phosphorus retention: Accounting for a missing piece in the watershed phosphorus puzzle.

Environmental Science and Technology **46**: 13284 – 13292.

- Javie, H.P., Sharpley, A.N., Withers, P.J.A., Scott, J.T., Haggard, B.E. and Neal, C. (2013) Phosphorus
   mitigation to control river eutrophication: Murky waters, inconvenient truths, and "postnormal"
   science. *Journal of Environmental Quality* 42: 295–304.
   Jensen, D.W., Steel E.A., Fullerton, A.H. and Pess, G.R. (2009) Impact of fine sediment on egg-to-fry
   survival of Pacific Salmon: A meta-analysis of published studies. Reviews in Fisheries Science 17:
- Jones, J.I., Murphy, J.F., Colins, A.L., Sear, D.A., Nadan, P.S. and Armitage, P.D. (2012) The impact of fine sediment on macroinvertebrates. River Research and Applications 28: 1055–1071.

348-359.

- Kefford, B.J., Zalizniak, L., Dunlop, J.E., Nugegoda, D. and Choy, S.C. (2010) How are
   macroinvertebrates of slow flowing lotic systems directly affected by suspended and deposited
   sediments? Environmental Pollution 158: 543–550.
- Kemp, P., Sear, D., Collins, A., Nadan, P. and Jones, I. (2011) The impact of fine sediment on riverine
   fish. Hydrological Processes 25: 1800–1821.
- Larsen, S., Pace, G. and Ormerod, S. J. (2011) Experimental effects of sediment deposition on the structure and function of macroinvertebrate assemblages in temperate streams. River Research and Applications 27: 257–267.
- Lindenmayer, D.B., Likens, G.E., Krebs, C.J. and Hobbs, R.J. (2010) Improved probability of detection
   of ecological "surprises". *Proceedings of the National Academy of Sciences of the United States of* America 107: 21957–21962.
- López-Rodríguez, M.J., de Figueroa, J.M.T. and Alba-Tercedor, J. (2009). The life history of *Serratella ignita* (Poda, 1761) (Insecta: Ephemeroptera) in a temporary and permanent Mediterranean stream. Aquatic Sciences-Research Across Boundaries 71: 179-188.
- 600 Lummer, E.M., Auerswald, K. and Geist, J. (2016) Fine sediment as environmental stressor affecting 601 freshwater mussel behavior and ecosystem services. *Science of the Total Environment* **571**: 1340-602 1348
- Macadam, C.R. and Bennett, C.J. (2010) A pictorial guide to the British Ephemeroptera. Field Studies
  Council. Shrewsbury, UK.
- Mainstone, C.P. and Parr, W. (2002) Phosphorus in rivers Ecology and management. Science of the Total Environment 282-283: 25–47.

607 Maturana, O. Tonina, D., McKean, J.A., Buffington, J.M., Luce, C.H. and Caamaño, D. (2014). 608 Modeling the effects of pulsed versus chronic sand inputs on salmonid spawning habitat in a low-609 gradient gravel-bed river. Earth Surface Processes and Landforms 39: 877–889. 610 Mantyka-Pringle, C.S., Martin, T.G., Moffatt, D.B., Linke, S. and Rhodes, J.R. (2014) Understanding 611 and predicting the combined effects of climate change and land-use changes on freshwater 612 macroinvertebrates and fish. Journal of Applied Ecology 51: 572-581. 613 Mathers, K.L., Rice, S.P. and Wood, P.J. (2017) Temporal effects of enhanced fine sediment loading 614 on macroinvertebrate community structure and functional traits. Science of the Total 615 Environment **599-600**: 513-522. 616 Meng, J., Yao, P., Yu, Z., Bianchi, T.S., Zhao, B., Pan, H. and Li, D. (2014) Speciation, bioavailability and 617 preservation of phosphorus in surface sediments of the Changjiang Estuary and adjacent East 618 China Sea inner shelf. Estuarine, Coastal and Shelf Science **144**: 27–38. 619 Minutoli, R., Granata, A. and Guglielmo, L. (2013) Potential use of ecotoxicological biomarkers in 620 Serratella ignita (Ephemeroptera) larvae for Alcantara river (Sicily, Italy) water quality 621 assessment. Journal of Limnology 72: 394-399. 622 Momoyama, K. (1987) Distributions of the hyphae in kuruma shrimps. Penaeus japonicas, infected 623 with Fusarium solani. Fish Pathology 22: 15-23. 624 Nijboer, R.C. and Verdonschot, P.F. (2004). Variable selection for modelling effects of eutrophication 625 on stream and river ecosystems. Ecological Modelling 177: 17–39. 626 Ortiz, J.D. and Puig, M.A. (2007) Point source effects on density, biomass and diversity of benthic 627 macroinvertebrates in a Mediterranean stream. River Research and Applications 23: 155–170. 628 Outram, F.N., Lloyd, C.E.M., Jonczyk, J., Benskin, C.McW.H., Grant, F., Perks, M.T., Deasy, C., Burke, 629 S.P., Collins, A.L., Freer, J., Haygarth, P.M., Hiscock, K.M., Johnes, P.J. and Lovett, A.L. (2014) 630 High-frequency monitoring of nitrogen and phosphorus response in three rural catchments to the 631 end of the 2011 – 2012 drought in England. Hydrology and Earth System Science 18: 3429 – 3448 632 Piggott, J.J., Lange, K., Townsend, C.R. and Matthaei, C.D. (2012) Multiple stressors in agricultural 633 streams: A mesocosm study of interactions among raised water temperature sediment addition 634 and nutrient enrichments. PLoS ONE 7: e49873. doi:10.1371/journal.pone.0049873

- Piggott, J.J., Townsend, C.R. and Matthaei, C.D. (2015) Climate warming and agricultural stressors
- interact to detrmine stream macroinvertebrate community dynamics. Global Change Biology 21:
- 637 1887–1906.
- 638 Pritchard, G., Harder, L.D. and Mutch, R.A. (1996) Development of aquatic insect eggs in relation to
- temperature and strategies for dealing with different thermal environments. *Biological Journal of*
- 640 the Linnean Society **58**: 221 244.
- 641 Przeslawski, R., Byrne, M. and Mellin, C. (2015) A review and meta-analysis of the effects of multiple
- abiotic stressors on marine embryos and larvae. Global Change Biology 21: 2122–2140.
- 643 Quinn, J.M., Davies-Colley, R.J., Hickey, C.W., Vickers, M.L. and Ryan, P.A. (1992) Effects of clay
- discharges on streams. Hydrobiologia 248: 235-247.
- 645 Reynolds, J.D. and Guillaume, H.P. (1998) Effects of phosphate on the reproductive symbiosis
- between bitterling and freshwater mussels: implications for conservation. Journal of Applied
- 647 *Ecology* **35**: 575 581.
- Rosenberg, D.M. and Wiens, A.P. (1978). Effects of sedimentation on microbenthic invertebrates on
- a northern Canadian river. Water Research 12: 753-763.
- Rotvit, L. and Jacobsen, D. (2013) Egg development of Plecoptera, Ephemeroptera and Odonata
- along latitudinal gradients. *Ecological Entomology* **39**: 177 185.
- Roy, A.H., A.D. Rosemond, D.S. Leigh, M.J. Paul and J.B. Wallace. (2003) Habitat specific responses of
- stream insects to land cover disturbance: biological consequences and monitoring implications.
- Journal of the North American Benthological Society 22: 292-307.
- Rutherford, R.B., Foster, B.L, Bammler, T., Beyer, R.P., Sato, S. and Somerman, M.J. (2006)
- 656 Extracellular Phosphate Alters Cementoblast Gene Expression. Journal of Dental Research 85:
- 657 505-509.
- 658 Sear, D.A., Jones, J.I., Collins, A.L., Hulin, A., Burke, N., Bateman, S., Pattison, I. and Nadan, P.S.
- 659 (2016) Does fine sediment source as well as quantity affect salmonid embryo mortality and
- development? *Science of the Total Environment* **541**: 957 968.
- 661 Sebater, S., Barcelo, D., De Castro-Català, N., Ginebreda, A., Kuzmanovic, M., Petrovic, M., Picó, Y.,
- 662 Ponsatí, L., Tornés, E. and Muñoz, I. (2016) Shared effects of organic microcontaminants and

- 663 environmental stressors on biofilms and invertebrates in impaired rivers. *Environmental Pollution*
- **210**: 308–314.
- Smith, V.H. and Schindler, D.W. (2009) Eutrophication science: where do we go from here? Trends in
- 666 Ecology & Evolution 24: 201–207.
- Sternecker, K. and Geist, J. (2010) The effects of stream substratum composition on the emergence
- of salmonid fry. *Ecology of Freshwater Fish* **19**: 537 544.
- 669 Strokal, M., Ma, L., Bai, Z. Luan, S. and Kroeze, C., Oenema, O., Velthof, G. and Zhang, F. (2016)
- Alarming nutrient pollution of Chinese rivers as a result of agricultural transitions. Environmental
- 671 Research Letters 11: 024014
- Tabak, L.M. and Gibbs, E. (1991). Effects of aluminium, calcium and low pH on egg hatching and
- 673 nymphal survival of Cloeon traingulifer McDunnough (Ephemeroptera: Baetidae). Hydrobiologia
- 674 218: 157-166.
- Tockner, K., Pusch, M., Borchardt, D. and Lorang, M.S. (2010) Multiple stressors in coupled river-
- floodplain ecosystems. *Freshwater Biology* **55**: 135–151.
- Tysmans, D.J.J., Löhr, A.J., Kroeze, C., Ivens, W.P.M.F. and van Wijnen, J. (2013) Spatial and temporal
- 678 variability of nutrient retention in river basins: A global inventory. *Ecological Indicators* **34**: 607 –
- 679 615.
- 680 US Environmental Protection Agency (2000). National water-quality inventory. Office of Water, US
- Environmental Protection Agency, Washington, DC.
- Voelkl, J., Alzoubi, K., Mamar, A. K., Ahmed, M. S. E., Abed, M. and Lang, F. (2014) Stimulation of
- 683 Suicidal Erythrocyte Death by Increased Extracellular Phosphate Concentrations. Kidney Blood
- Pressure Research 38: 42-51.
- von Bertrub, M.G., Krein, A., Stendera, S., Thielen, F. and Hering, D. (2013) Is fine sediment
- deposition a main driver for the composition of benthic macroinvertebrate assemblages?
- 687 *Ecological Indicators* **24**: 589–598.
- Wagenhoff, A., Townsend, C.R. & Matthaei, C.D. (2012) Macroinvertebrate responses along broad
- stressor gradients of deposited fine sediment and dissolved nutrients: a stream mesocosm
- experiment. Journal of Applied Ecology 49: 892–902.

691 Wedekend, C, Gessner, M.O., Vazquez, F., Maerki, M. and Steiner, D. (2010) Elevated resource 692 availability sufficient to turn opportunistic into virulent fish pathogens. Ecology 91: 1251-1256. 693 Wood, P.J. and Armitage, P.D. (1997) Biological effects of fine sediment in the lotic environment. 694 Environmental Management 21: 203–217. 695 Wood, S.A., Depree, C., Brown, L., McAllister, T. and Hawes, I. (2015) Entrapped sediments as a 696 source of phosphorus in epilithic cyanobacterial proliferations in low nutrient rivers. PLoS ONE 697 10: e0141063. https://doi.org/10.1371/journal.pone.0141063 698 Worrell, F., Jarvie, H.P., Howden, N.J.K. and Burt, T.P. (2016) The fluvial flux of total reactive and 699 total phosphorus from the UK in the context of a national phosphorus budget: comparing UK 700 river fluxes with phosphorus trade imports and exports. Biogeochemistry 130: 31-51. 701 WWF-UK (2007). Review of UKTAG Proposed Standard for Suspended Solids. APEM Scientific Report, 702 20pp. 703 Yan, Z., Han, W., Peñuelas, J., Sardans, J., Elser, J.J., Du, E., Reich, P.B. and Frang, J. (2016) 704 Phosphorus accumulates faster than nitrogen globally in freshwater ecosystems under 705 anthropogenic impacts. Ecology Letters 19: 1237 – 1246. 706 Yeo, R.R. and Dechoretz, N. (1976) Diquat and Copper-Ion Residues in Salmon-Spawning Channel. 707 Weed Science, 405-409. 708 Zhao, Y. and Newman, M.C. (2006) Effects of exposure duration and recovery time during pulsed 709 exposures. *Environmental Toxicology and Chemistry* **25**: 1298–1304.

**Table 1:** Experimental designs for bioassays.

| Chemical treatment              | Nominal chemical concentration mg/l |      |      |     |     |     |  |
|---------------------------------|-------------------------------------|------|------|-----|-----|-----|--|
| Fine suspended solids (SS)      | 0                                   | 5    | 10   | 15  | 20  | 25  |  |
| Orthophosphate (OP)             | 0                                   | 0.05 | 0.07 | 0.1 | 0.2 | 0.3 |  |
| 0.07 mg l <sup>-1</sup> OP + SS | 0                                   | 5    | 10   | 15  | 20  | 25  |  |
| 10 mg l <sup>-1</sup> SS + OP   | 0                                   | 0.05 | 0.07 | 0.1 | 0.2 | 0.3 |  |

 Table 2: Mean physical-chemical properties for each bioassay.

| Parameter                                      | Mean water quality ± s.d. (n = 10) |                 |                 |                  |                 |                |
|--|------------------------------------|-----------------|-----------------|------------------|-----------------|----------------|
| Nominal SS concentration (mg l-1)              | 0                                  | 5               | 10              | 15               | 20              | 25             |
| Actual SS concentration (mg l-1)               | <3                                 | $5.2 \pm 0.2$   | $10.0 \pm 0.1$  | $14.9 \pm 0.2$   | $20.0 \pm 0.2$  | $24.7 \pm 0.6$ |
| Water temperature (°C)                         | 13.4 ± 4.9                         | 13.4 ± 4.9      | 13.5 ±4.9       | $13.4 \pm 5.0$   | $13.3 \pm 5.0$  | 13.3 ± 4.9     |
| Dissolved oxygen (mg l-1)                      | $10.5 \pm 0.4$                     | $10.3 \pm 0.4$  | $10.3 \pm 0.4$  | $10.3 \pm 0.4$   | 10.2 ±0.2       | $10.3 \pm 0.4$ |
| рН   | 8.0 ± 0.1                          | $8.0 \pm 0.1$   | $8.0 \pm 0.1$   | $7.9 \pm 0.3$    | $7.9 \pm 0.1$   | 7.9 ± 0.1      |
| Nominal OP concentration (mg l <sup>-1</sup> ) | 0                                  | 0.05            | 0.07            | 0.1              | 0.2             | 0.3            |
| Actual OP concentration (mg l <sup>-1</sup> )  | 0.04 ±<br>0.01                     | $0.05 \pm 0.01$ | $0.07 \pm 0.02$ | $0.11 \pm 0.01$  | $0.20 \pm 0.01$ | 0.30 ±<br>0.01 |
| Water temperature (°C)                         | 13.4 ± 5.1                         | 13.1 ± 4.9      | $13.0 \pm 5.0$  | 13.0 ± 4.9       | $12.9 \pm 5.0$  | $12.8 \pm 5.0$ |
| Dissolved oxygen (mg l-1)                      | 10.5 ± 0.4                         | $10.2 \pm 0.3$  | $10.2 \pm 0.3$  | $10.3 \pm 0.3$   | $10.2 \pm 0.3$  | $10.3 \pm 0.4$ |
| рН   | 7.9 ± 0.1                          | $7.9 \pm 0.1$   | $7.9 \pm 0.1$   | $7.8 \pm 0.3$    | $7.8 \pm 0.1$   | $7.8 \pm 0.1$  |
| Nominal SS concentration (mg l-1)              | 0                                  | 5               | 10              | 15               | 20              | 25             |
| Actual SS concentration (mg l-1)               | <3                                 | $5.2 \pm 0.4$   | $10.0 \pm 0.1$  | $15.0 \pm 0.1$   | $20.1 \pm 0.2$  | $25.0 \pm 0.2$ |
| Nominal OP concentration (mg l-1)              | 0.07                               | 0.07            | 0.07            | 0.07             | 0.07            | 0.07           |
| Actual OP concentration (mg l-1)               | 0.04 ±<br>0.01                     | $0.08 \pm 0.01$ | $0.07 \pm 0.01$ | $0.07 \pm 0.004$ | 0.07 ±0 .003    | 0.07 ±<br>0.01 |
| Water temperature (°C)                         | 13.7 ±5.2                          | $13.6 \pm 4.7$  | 13.6 ± 4.5      | $13.3 \pm 4.9$   | $13.0 \pm 4.7$  | 12.9 ± 4.8     |
| Dissolved oxygen (mg l-1)                      | 10.4 ±0.4                          | $10.0 \pm 0.5$  | $10.2 \pm 0.3$  | $10.2 \pm 0.3$   | $10.2 \pm 0.4$  | $10.0 \pm 0.3$ |
| рН   | 8.0 ± 0.1                          | $8.1 \pm 0.1$   | $8.0 \pm 0.1$   | $8.0 \pm 0.3$    | $8.0 \pm 0.1$   | $8.2 \pm 0.2$  |
| Nominal OP concentration (mg l <sup>-1</sup> ) | 0                                  | 0.05            | 0.07            | 0.1              | 0.2             | 0.3            |
| Actual OP concentration (mg l-1)               | 0.04 ±                             | $0.05 \pm$      | $0.07 \pm$      | $0.10 \pm 0.01$  | $0.21 \pm 0.01$ | $0.31 \pm$     |
|  | 0.01                               | 0.004           | 0.004           |                  |                 | 0.01           |
| Nominal SS concentration (mg l-1)              | 10                                 | 10              | 10              | 10               | 10              | 10             |
| Actual SS concentration (mg l-1)               | <3                                 | $9.9 \pm 0.5$   | $10.0 \pm 0.2$  | $10.2 \pm 0.3$   | $9.9 \pm 0.3$   | $10.0 \pm 0.1$ |
| Water temperature (°C)                         | 13.3 ± 4.6                         | $13.3 \pm 4.7$  | $13.4 \pm 5.2$  | $13.4 \pm 4.7$   | $13.1 \pm 5.0$  | $13.0 \pm 4.9$ |
| Dissolved oxygen (mg l-1)                      | $10.2 \pm 0.4$                     | $10.7 \pm 0.4$  | $11.0 \pm 0.6$  | $10.3 \pm 0.5$   | $10.1 \pm 0.4$  | $10.1 \pm 0.3$ |
| рН   | $8.0 \pm 0.1$                      | 8.1 ± 0.1       | $8.0 \pm 0.2$   | $7.9 \pm 0.3$    | $8.3 \pm 0.2$   | $8.3 \pm 0.3$  |

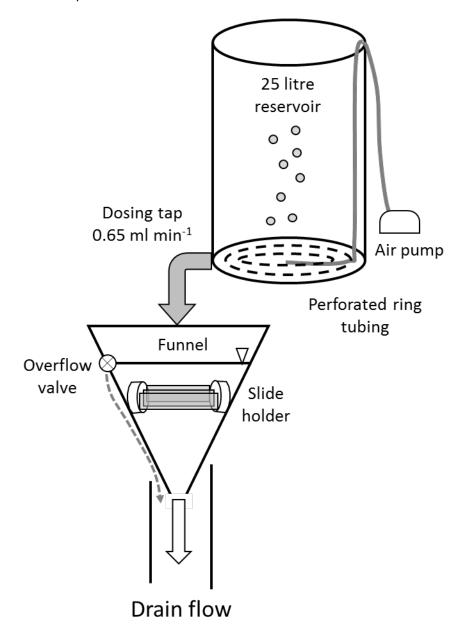
**Table 3:** Regression equations and significance values for OP and/or SS concentration against egg mortality. Note relationships are all exponential with the exception of OP + SS, where the strongest relationship was linear. All regressions are significant (p < 0.01).

| Treatment | Time (days) | Equation                  | R <sup>2</sup> |
|-----------|-------------|---------------------------|----------------|
| OP        | 72          | 25.495e <sup>5.265x</sup> | 0.88           |
|           | 121         | 29.162e <sup>7.225x</sup> | 0.98           |
|           | 183         | 29.324e <sup>8.189x</sup> | 0.98           |
| SS        | 72          | 22.817e <sup>0.099x</sup> | 0.99           |
|           | 121         | 25.832e <sup>0.115x</sup> | 0.97           |
|           | 183         | 27.70e <sup>0.120x</sup>  | 0.98           |
| OP + SS   | 72          | 781.39x + 5.8258          | 0.98           |
|           | 121         | 921.45x + 35.319          | 0.92           |
|           | 183         | 1155.2x + 51.769          | 0.93           |
| SS + OP   | 72          | 22.446e <sup>0.095x</sup> | 0.96           |
|           | 121         | 27.646e <sup>0.113x</sup> | 0.98           |
|           | 183         | 36.937e <sup>0.111x</sup> | 0.98           |

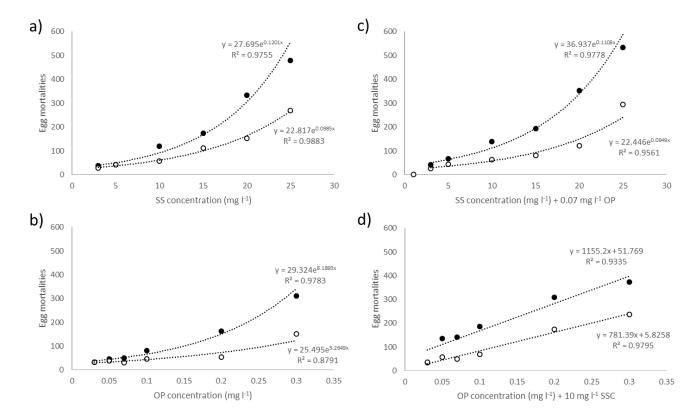
**Table 4:** Analysis of national routine spot measures of OP and SS made by the Environment Agency
 724 in 2015 for WFD compliance across England.

|                   |    | Count | Average<br>(mg l <sup>-1</sup> ) | Percentage of sites where the average is       |                          |   |                          |  |  |
|-------------------|----|-------|----------------------------------|--|--------------------------|---|--------------------------|--|--|
|                   |    |       | (IIIg I -)                       | > 0.3 mg l <sup>-1</sup>                       | > 0.1 mg l <sup>-1</sup> | 9.2  where every meas  1 < 0.1 mg l-1  21.8  where every meas | > 10 mg l <sup>-1</sup>  |  |  |
| All sites         | OP | 32549 | 0.27                             | 21.8   | 51.6                     |   |                          |  |  |
| All sites         | SS | 2029  | 16.6                             |  |                          | 9.2   | 31.6                     |  |  |
|                   |    | Count | Average                          | Percent  | age of sites who         | ere every meas  | urement is               |  |  |
|                   |    |       | (mg l <sup>-1</sup> )            | > 0.3 mg l <sup>-1</sup>                       | > 0.1 mg l <sup>-1</sup> | < 0.1 mg l <sup>-1</sup>                                      | < 0.3 mg l <sup>-1</sup> |  |  |
| Sites > 9 samples | OP | 1812  | 0.30                             | 22.2   | 52.9                     | 21.8  | 52.7                     |  |  |
|                   |    | Count | Average                          | Percentage of sites where every measurement is |                          |   |                          |  |  |
|                   |    |       | (mg l-1)                         | > 25 mg l <sup>-1</sup>                        | > 10 mg l <sup>-1</sup>  | < 10 mg l <sup>-1</sup>                                       | < 25 mg l <sup>-1</sup>  |  |  |
| Sites > 9 samples | SS | 129   | 12.7                             | 9.4  | 39.0                     | 49.6  | 10.9                     |  |  |

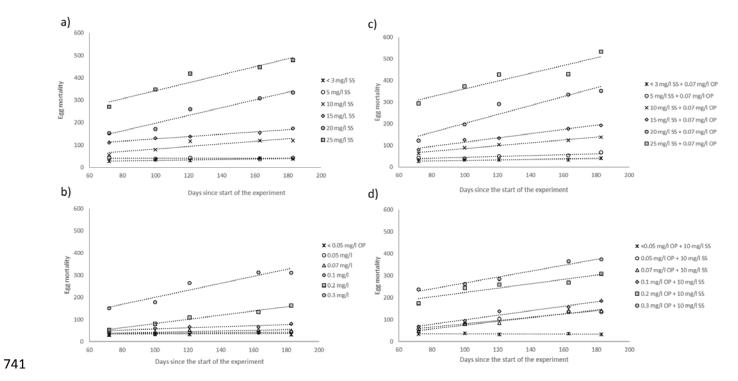
**Figure 1:** Schematic of *S. ignita* egg dosing rigs for controls and treatments. A reservoir with experiment solution is held above a funnel, within which 3 slides containing *S. ignita* eggs are held in a slide holder. A perforated ring of tubing on the base of the reservoir ensured complete mixing and aeration of experimental water.



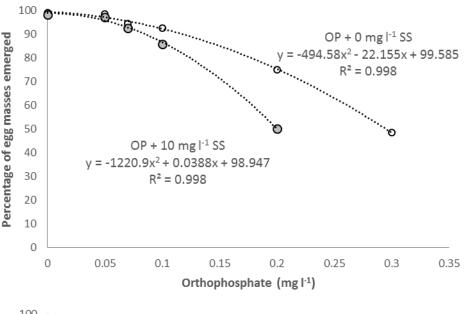
**Figure 2:** Regressions of the mortality of *S.ignita* eggs after 72 days exposure (open circles) and 183 days exposure (filled circles) against (a) SS, (b) OP), (c) SS in addition to 0.07 mg  $l^{-1}$  OP, and (d) OP in addition to 10 mg  $l^{-1}$  SS.

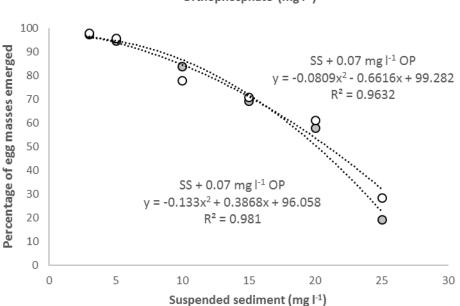


# **Figure 3:** Mortality of *S. ignita* eggs through time under differing concentrations of (a) SS; (b) OP; (c) SS plus 0.07 mg $I^{-1}$ OP, and; (d) OP plus 10 mg $I^{-1}$ SS.

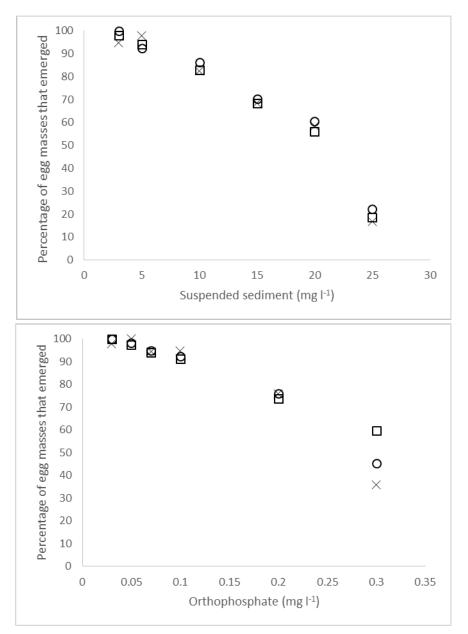


**Figure 4**: Percentage of *S.ignita* egg masses surviving to emergence under different concentrations of (a) OP in isolation (open circles) and in combination with 10 mg  $I^{-1}$  of SS (closed circles) and (b) SS in isolation (open circles) and in combination with 0.07 mg  $I^{-1}$  of OP. Note, at 0.3 mg  $I^{-1}$  OP plus 10 mg  $I^{-1}$  SS, fungal growth prevented the majority of egg masses from emerging and prevented an accurate count of egg mass emergence.





**Figure 5:** Percentage of *S.ignita* egg masses surviving to emergence under different concentrations of (a) OP (b) SS for each of the 3 slides in each treatment indicated separately.



# SUPPLEMENTARY MATERIAL A:

758 Image of spent *Serratella ignita* with eggs deposited on glass slides at the bottom of the tray.



# SUPPLEMENTARY MATERIAL B:

763 Image of dosing funnel chamber containing glass slides with deposited egg masses.



# SUPPLEMENTARY MATERIAL C:

Image of displaced or 'ghost' S.ignita egg masses.



# **SUPPLEMENTARY MATERIAL D:**

# 771 Mean physical-chemical properties for control and diluent carbon-limestone filtered tap water.

| Parameter                        | Unit               | Sample date (n = 8) |         |          |          |          |         |         |         |
|----------------------------------|--------------------|---------------------|---------|----------|----------|----------|---------|---------|---------|
|                                  |                    | 29/8/15             | 26/9/15 | 31/10/15 | 28/11/15 | 19/12/15 | 30/1/16 | 19/2/16 | 26/3/16 |
| Nitrogen: Total as<br>N          | mg l <sup>-1</sup> | 0.5                 | 0.525   | 0.435    | 0.44     | 0.535    | 0.501   | 0.702   | 0.563   |
| Alkalinity to pH<br>4.5 as CaCO3 | mg l <sup>-1</sup> | 117                 | 122     | 129      | 139      | 144      | 154     | 150     | 161     |
| Ammoniacal<br>Nitrogen as N      | mg l <sup>-1</sup> | <0.030              | <0.030  | <0.030   | <0.0300  | <0.0300  | <0.0300 | <0.0300 | <0.0300 |
| Nitrite as N                     | mg l <sup>-1</sup> | <0.0040             | <0.0040 | <0.0040  | <0.0040  | <0.0040  | <0.0040 | <0.0040 | <0.0040 |
| Orthophosphate as P              | mg l <sup>-1</sup> | 0.044               | 0.045   | 0.027    | 0.033    | 0.03     | 0.036   | 0.04    | 0.029   |
| рН                               |                    | 7.97                | 8.03    | 7.92     | 7.99     | 8.07     | 8.07    | 8.14    | 8.27    |
| Suspended Solids<br>at 105° C    | mg l <sup>-1</sup> | <3                  | <3      | <3       | <3       | <3       | <3      | <3      | <3      |
| Boron                            | μg l <sup>-1</sup> | <100                | <100    | <100     | <100     | <100     | <100    | <100    | <100    |
| Calcium                          | mg l <sup>-1</sup> | 49.2                | 58.1    | 49       | 58       | 59.3     | 61.9    | 66.9    | 65.1    |
| Iron                             | μg l <sup>-1</sup> | 34.9                | 49.9    | 45.2     | 37.6     | 39.9     | 32.8    | 40.9    | 34.7    |
| Lithium                          | μg l <sup>-1</sup> | <100                | <100    | <100     | <100     | <100     | <100    | <100    | <100    |
| Magnesium                        | mg l <sup>-1</sup> | 2.55                | 3.46    | 5.02     | 3.05     | 3.91     | 3.32    | 3.37    | 3.86    |
| Manganese                        | μg l <sup>-1</sup> | <10                 | <10     | <10      | <10      | <10      | <10     | <10     | <10     |
| Sodium                           | mg l <sup>-1</sup> | 8.19                | 9.1     | 8.65     | 8.17     | 8.28     | 8.4     | 8.34    | 8.35    |
| Mercury                          | μg l <sup>-1</sup> | <0.01               | <0.01   | <0.01    | <0.01    | <0.01    | <0.01   | <0.01   | <0.01   |
| Hardness : Total<br>as CaCO3     | mg l <sup>-1</sup> | 136                 | 159     | 143      | 157      | 164      | 168     | 181     | 178     |
| Arsenic                          | μg l <sup>-1</sup> | 1                   | -       | -        | _        | -        | -       | -       | 1       |
| Selenium                         | μg l <sup>-1</sup> | 1                   | -       | -        | -        | -        | -       | -       | 1       |
| Cadmium                          | μg l <sup>-1</sup> | 0.1                 | -       | -        | -        | -        | -       | -       | 0.2     |
| Copper                           | μg l <sup>-1</sup> | 0.5                 | -       | -        | -        | -        | -       | -       | 0.115   |
| Lead                             | μg l <sup>-1</sup> | 0.61                | -       | -        | -        | -        | -       | -       | 0.5     |
| Nickel                           | μg l <sup>-1</sup> | 1.97                | -       | _        | -        | -        | -       | _       | 0.4     |

# **SUPPLEMENTARY MATERIAL E:**

# 777 Water temperature in a control *S. ignita* egg chamber.

